Bioengineering 280A Principles of Biomedical Imaging

> Fall Quarter 2008 MRI Lecture 1

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Spin

- Intrinsic angular momentum of elementary particles -- electrons, protons, neutrons.
- Spin is quantized. Key concept in Quantum Mechanics.

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Topics

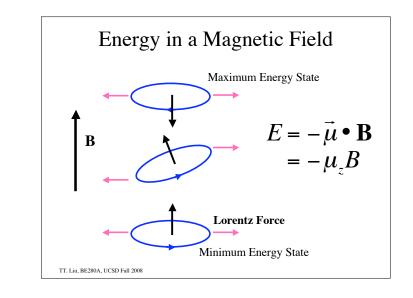
- The concept of spin
- Precession of magnetic spin
- Relaxation

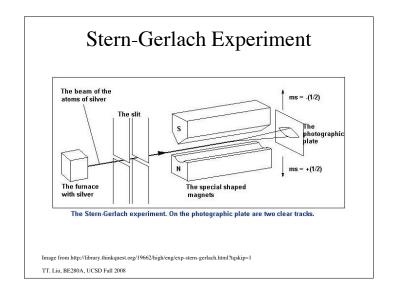
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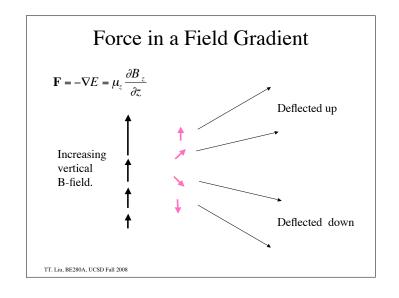
The History of Spin

- 1921 Stern and Gerlach observed quantization of magnetic moments of silver atoms
- 1925 Uhlenbeck and Goudsmit introduce the concept of spin for electrons.
- 1933 Stern and Gerlach measure the effect of nuclear spin.
- 1937 Rabi predicts and observes nuclear magnetic resonance.

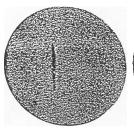
Classical Magnetic Moment $\vec{\mu} = IA\hat{n}$

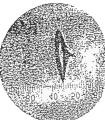






Stern-Gerlach Experiment





 $Image\ from\ http://library.thinkquest.org/19662/high/eng/exp-stern-gerlach.html?tqskip=1$

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Magnetic Moment and Angular Momentum



A charged sphere spinning about its axis has angular momentum and a magnetic moment.

This is a classical analogy that is useful for understanding quantum spin, but remember that it is only an analogy!

Relation: $\mu = \gamma \, S$ where γ is the gyromagnetic ratio and S is the spin angular momentum.

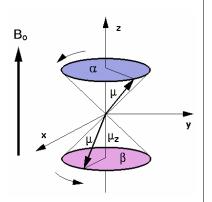
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Quantization of Magnetic Moment

The key finding of the Stern-Gerlach experiment is that the magnetic moment is quantized. That is, it can only take on discrete values.

In the experiment, the finding was that the component of magnetization along the direction of the applied field was quantized:

$$\mu_z = + \mu_0 \text{ OR} - \mu_0$$



http://www.le.ac.uk/biochem/mp84/teaching

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Quantization of Angular Momentum

Because the magnetic moment is quantized, so is the angular momentum.

In particular, the z-component of the angular momentum Is quantized as follows:

$$S_{\cdot} = m_{s}\hbar$$

$$m_s \in \{-s, -(s-1), \dots s\}$$

s is an integer or half intege

Nuclear Spin Rules

Number of Protons	Number of Neutrons	Spin	Examples
Even	Even	0	¹² C, ¹⁶ O
Even	Odd	j/2	¹⁷ O
Odd	Even	j/2	¹ H, ²³ Na, ³¹ P
Odd	Odd	j	² H

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Magnetic Field Units 1 Tesla = 10,000 Gauss Earth's field is about 0.5 Gauss 0.5 Gauss = 0.5×10^{-4} T = 50μ T

Hydrogen Proton

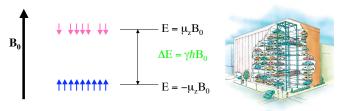
Spin 1/2

$$S_z = \begin{cases} +\hbar/2 \\ -\hbar/2 \end{cases}$$

$$\mu_z = \begin{cases} +\gamma \hbar/2 \\ -\gamma \hbar/2 \end{cases}$$

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Boltzmann Distribution



 $\frac{\text{Number Spins Up}}{\text{Number Spins Down}} = \exp(-\Delta E/kT)$

Ratio = 0.999990 at 1.5T !!!
Corresponds to an excess of about 10 up spins per million

Equilibrium Magnetization

$$\begin{aligned} \mathbf{M}_{0} &= N \left\langle \boldsymbol{\mu}_{z} \right\rangle = N \left(\frac{n_{up} \left(\boldsymbol{\mu}_{z} \right) - n_{down} \left(\boldsymbol{\mu}_{z} \right)}{N} \right) \\ &= N \mu \frac{e^{\mu_{z} B / (kT)} - e^{-\mu_{z} B / (kT)}}{e^{\mu_{z} B / (kT)} + e^{-\mu_{z} B / (kT)}} \\ &\approx N \mu_{z}^{2} B / (kT) \\ &= N \gamma^{2} \hbar^{2} B / (4kT) \end{aligned}$$

N = number of nuclear spins per unit volume Magnetization is proportional to applied field.

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Gyromagnetic Ratios

Nucleus	Spin	Magnetic Moment	$\gamma/(2\pi)$ (MHz/Tesla)	Abundance
¹H	1/2	2.793	42.58	88 M
²³ Na	3/2	2.216	11.27	80 mM
³¹ P	1/2	1.131	17.25	75 mM

Source: Haacke et al., p. 27

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Bigger is better



3T Human imager at UCSD.



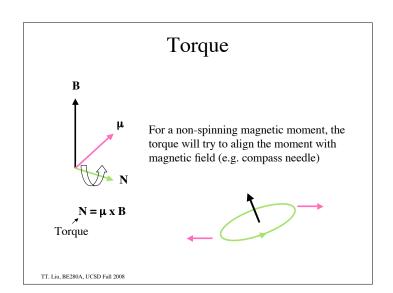
7T Rodent Imager at UCSD

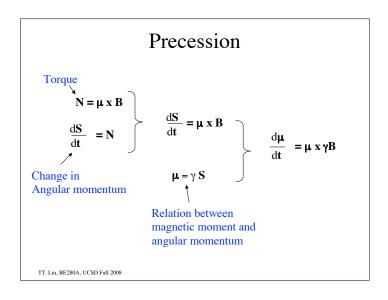


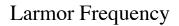
7T Human imager at U. Minn.



9.4T Human imager at UIC







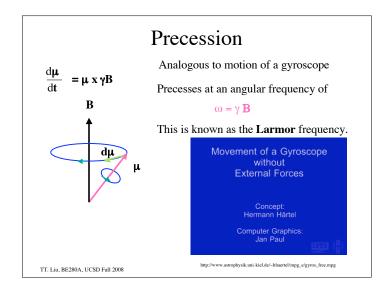
 $\omega = \gamma B$ Angular frequency in rad/sec

 $f = \gamma B / (2 π)$ Frequency in cycles/sec or Hertz, Abbreviated Hz

For a 1.5 T system, the Larmor frequency is 63.86 MHz which is 63.86 million cycles per second. For comparison, KPBS-FM transmits at 89.5 MHz.

Note that the earth's magnetic field is about 50 μ T, so that a 1.5T system is about 30,000 times stronger.

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Notation and Units

1 Tesla = 10,000 Gauss

Earth's field is about 0.5 Gauss

 $0.5 \text{ Gauss} = 0.5 \times 10^{-4} \text{ T} = 50 \,\mu\text{T}$

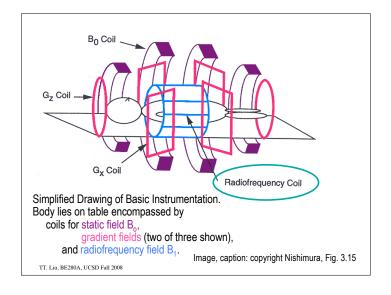
 $\gamma = 26752 \text{ radians/second/Gauss}$

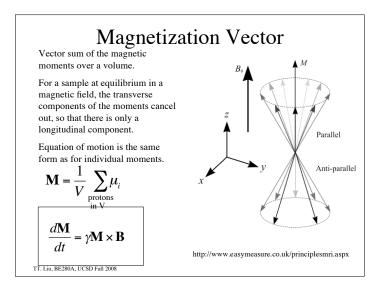
 $\gamma = \gamma/2\pi = 4258$ Hz/Gauss

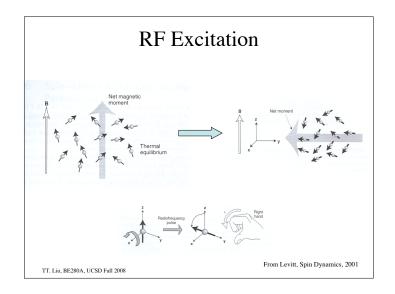
= 42.58 MHz/Tesla

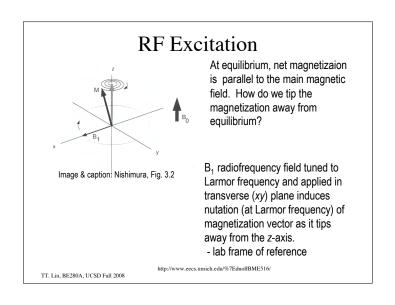
Recap

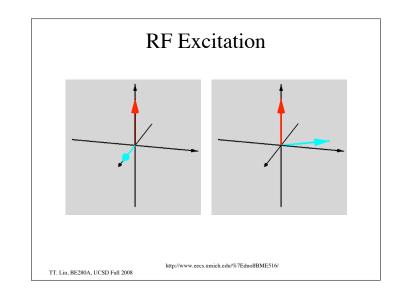
- Spins: angular momentum and magnetic moment are quantized.
- Spins precess about a static field at the Larmor frequency.
- In MRI we work with the net magnetic moment.
- In the presence of a static field and non-zero temperature, the equilibirum net magnetic moment is aligned with the field (longitudinal), since transverse components cancel out.
- We will use an radiofrequency pulse to tip this longitudinal component into the transverse plane.

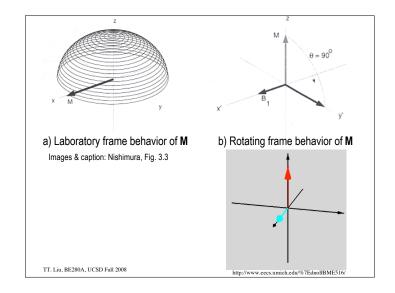


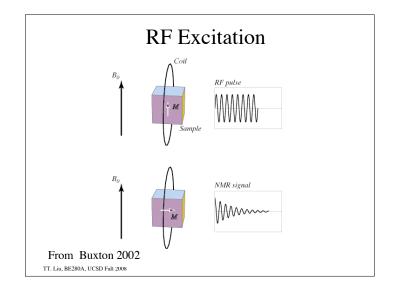


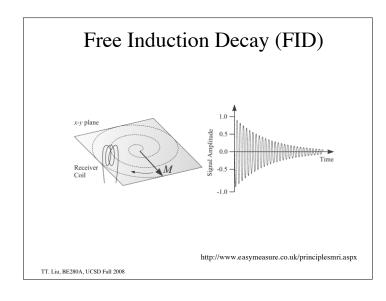














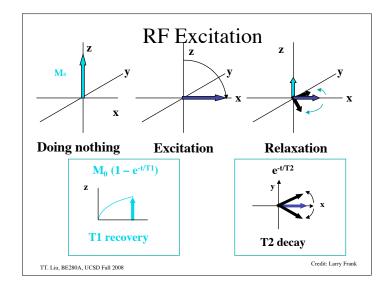
An excitation pulse rotates the magnetization vector away from its equilibrium state (purely longitudinal). The resulting vector has both longitudinal $\mathbf{M_z}$ and tranverse $\mathbf{M_{xy}}$ components.

Due to thermal interactions, the magnetization will return to its equilibrium state with characteristic time constants.

 T_1 spin-lattice time constant, return to equilibrium of M_z

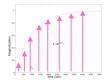
 T_2 spin-spin time constant, return to equilibrium of $\mathbf{M}_{\mathbf{x}\mathbf{y}}$

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Longitudinal Relaxation

$$\frac{d\mathbf{M}_z}{dt} = -\frac{M_z - M}{T_1}$$

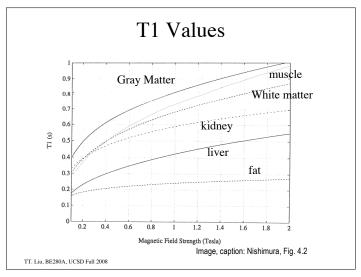


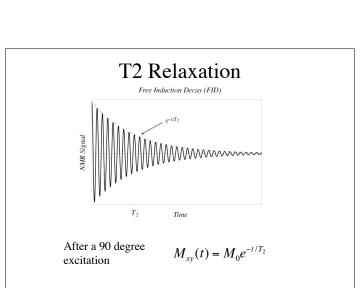
After a 90 degree pulse

$$M_{z}(t) = M_{0}(1 - e^{-t/T_{1}})$$

Due to exchange of energy between nuclei and the lattice (thermal vibrations). Process continues until thermal equilibrium as determined by Boltzmann statistics is obtained.

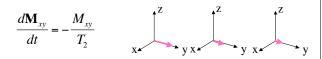
The energy ΔE required for transitions between down to up spins, increases with field strength, so that T_1 increases with **B**.





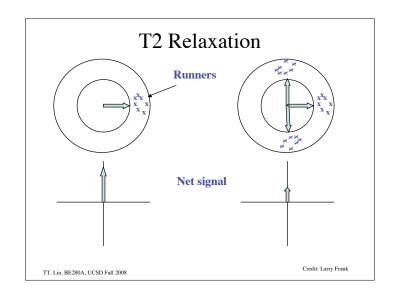
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Transverse Relaxation



Each spin's local field is affected by the z-component of the field due to other spins. Thus, the Larmor frequency of each spin will be slightly different. This leads to a dephasing of the transverse magnetization, which is characterized by an exponential decay.

 T_2 is largely independent of field. T_2 is short for low frequency fluctuations, such as those associated with slowly tumbling macromolecules.



T2 Values

Tissue	T ₂ (ms)	
gray matter	100	
white matter	92	
muscle	47	
fat	85	
kidney	58	
liver	43	
CSF	4000	

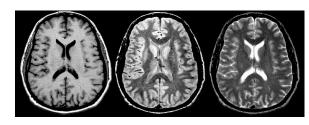
Solids exhibit very short T₂ relaxation times because there are many low frequency interactions between the immobile spins.

On the other hand, liquids show relatively long T_2 values, because the spins are highly mobile and net fields average out.

Table: adapted from Nishimura, Table 4.2

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Example



T₁-weighted

Density-weighted

T₂-weighted

Questions: How can one achieve T2 weighting? What are the relative T2's of the various tissues?